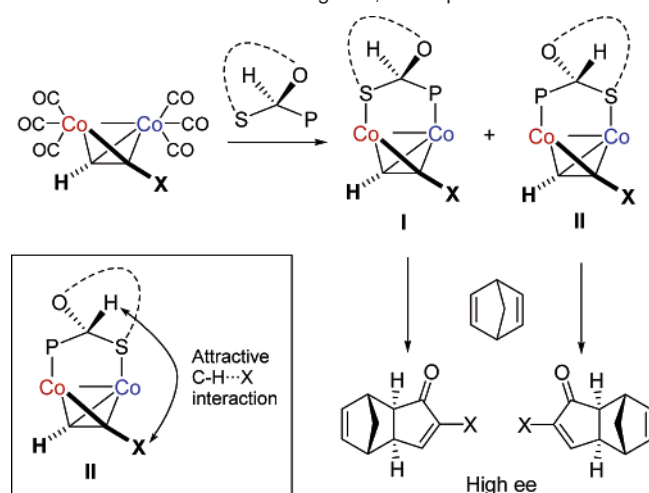


Phosphine–Substrate Recognition through the C–H···O Hydrogen Bond: Application to the Asymmetric Pauson–Khand Reaction [*J. Am. Chem. Soc.* **2005**, *127*, 13629–13633]. Jordi Solà, Antoni Riera,* Xavier Verdagué,* and Miguel A. Maestro

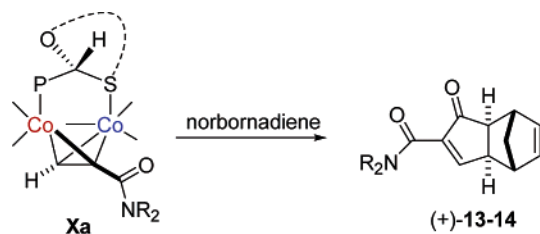
Pages 13630 and 13632. It has come to our attention that, in this report, the stereochemistry of the Pauson–Khand adducts depicted in Scheme 1 and Table 2 was shown incorrectly. The stereochemistry of the intermolecular adducts is correctly described in the Experimental Section. We regret these mistakes. The correct Scheme 1 and Table 2 graphics are presented below.

Scheme 1. Diastereomeric Bridged P,S Complexes^a



^a Carbonyl groups, phosphine substituents, and ligand backbone are omitted for clarity. Just one of the two ligand alignments grants a contact within the acceptor (X) and the methine group on the ligand.

Table 2. Intermolecular Pauson–Khand Reaction of Major Diastereomers



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The Interaction of Water with MOF-5 Simulated by Molecular Dynamics [*J. Am. Chem. Soc.* **2006**, *128*, 10678–10679]. Jeffery A. Greathouse* and Mark D. Allendorf

Supporting Information, Table SI-1. The values of the van der Waals well depth parameters (ϵ) given were those in units of kcal·mol⁻¹, although the heading showed kJ·mol⁻¹. The Supporting Information has been amended to show the correct values in units of kJ·mol⁻¹. The authors are grateful to David Dubbeldam for pointing out this error.

Supporting Information Available: Simulation methods and structural results. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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